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# U.S. DEPARTMENT OF AGRICULTURE. DIVISION OF CHEMISTRY. 

## REPOR'T

or
EXPLBRIUENTS IN TIIE MANUFAGTURE OF SUGAR

AT

MAGNOLIA STATION, LAWRENCE, LA.,

SEASON OF 1886-1887.

THIRD REPORT,
BY
GUILFORD L. SPENCER.

## LETTERS OF TRANSMIT'AL.

## United States Department of Agriculture, Washington, D. C., April 12, 1887.

SIR: I have the honor to submit herewith for your inspection and approval the results of the experimental work carried on by your direction in Louisiana during the manufacturmg season of 1886 by Mr. G. L. Spencer, assistant in the Division of Chemistry.

The information collected is of the highest interest to Louisiana sugar growers, and is made still more valuable by some comparative studies of the Cuban sugar industry.

Very respectfully,

H. W. WILEY,<br>Chemist.

> Hon. Norman J. Colman, Commissioner.

> United States Department of Agriculture, Division of Chemistry, Washington, D. U., March 31, 1887.

Sin: I have the honor to submit the third report of the Department's experimeut station at Magnolia plantation.

Included in this report is a brief statement of the work of three Cuban sugar-houses, indicating certain economies that it would be well for the planters of this country to adopt.

Accompanyiug this report is an article on sugar analysis, prepared at my request by Dr. C. A. Crampton. This contains a valuable table, devised by Dr. Crampton some years since, and which is now employed at the Department's stations for the calculation of analyses.

Very respectfully,

G. L. SPENCER, Assistant Chemist.

Dr. H. W. Wiley, Chemist.

## EXPERIMENTS IN THE MANUFACTURE OF SUGAR.

The manufacturing season at Magnolia commenced November 7. 1886, and ended December 20. This completes the thirl season's work of the 1) epartment at this station.

I shall give, in as few words as possible, a comparison of the growing seasons of the past three years.

In 1884 the weather was favorable until the 1st of June, then followed a period of very wet weather, lasting until August, which was a very dry month. The conditions in September and October were farorable to the ripening of the cane. During the rolling season there were frequent and heavy rains.

Season of 1885.-The early part of this season was exceptionally wet. From April to July the rainfall was limited to but three or four showers; in August and September the rains were frequent and heavy ; in October and the rest of the season the weather was exceptionally dry and cool, the mean temperature being considerably below that of 1884 . The cane was prostrated in September by a heavy wind storm.

Season of 1886.-In consequence of cold, wet weather in February, March, and April the cane obtained a very late start. May was dry and cool ; June and July were so wet that it was impossible to properiy cultivate the cane; August was dry and exceedingly hot. In September, October, and the rolling season the weather was very dry. The drought probably saved the cane from damage in the heavy wind storm of October, when the lower coast of the Mississippi near Pointe à la Hãche was seriously damaged. There was a killing freeze November 17. This is the earliest freeze, with one exception, noted in the plantation records, extending over a long period of years. December was cold.

It may be seen from the above statements that these three seasons differed very materially from one another. That of 1884 might be considered very favorable. The tonnage was fair and the cane rich. In 1885 the conditions were also favorable, with the exception of the wet weather in September and the heavy wind storm. The tonnage was large, but the quality of the cane was poor.

In January, 1886, there was a severe freeze, perhaps the most serere on record in Lonisiana. At the time of this freeze it was feared that
the stubble had been damaged by the frost, but such did not prove to be the case.

It is customary at Magnolia to burn the trasli in the fields soon after the cane is all harvested. There is a general impression among planters that this exposes the stribble to danger from frost. The experience of the past season at Magnolia demonstrates that such is not the case.

The tomage this season has been unusually small, but the cane has been a little richer in available sucrose than any time since the station was established.

In comparing the work of the sugar-house it is probably better to compare this season's work with that of 1884-85. The same grades of sugar were made these seasons.

The yield per acre in 1885-'86 was 2,988 pounds of first, second, and third sugars. In 1886 the field, first and second sugars, was 1,964 pounds. A decrease of 6.66 tons of cane per acre resulted in a decrease of 1,024 pounds of sugar.

The yield of sugar per ton of cane the past season has been exceedingly satisfactory, but nevertheless it is expected that a few changes in the house will materially increase the output of sugar.

This season ouly two sugars were made. The first graded "Clioice off white." The sield of first sugar would have been larger had it been possible to boil the masse cuite stiffer, but the strike ralve of the pan would not admit of doing so. As large a proportion of the sugar as possible should be obtained in the firsts, not only on account of the higher price of this grade of sugar, but because less sugar is left to be reboiled for lower grades and consequently the loss from inversion is diminished.

Since most of the sugars and molasses in Louisiana are made to enter directly into consumption, the planter is compelled to lime the juice lower and boil hotter than he would otherwise. Were it practicable by present methods to work alkaline the loss from inversion even at high temperatures would be greatly diminished.

I have examined beet molasses from sugars boiled with a racum of less than 24 inches that contained but little more than a trace of invert sugar. The clarified juices and sirups were always worked alkaline.

IMPROVEMENTS IN THE SUGAR-HOUSE AND THE PLANTATION.
Very few changes were made in the house preparatory to this season's work, and those only for the improvement of the work of machinery already in place.

The tanks for skimmings were provided with a better arrangement for decanting the clear juice. The lowest ontlets from these tanks are about 2 inches above the bottoms. Ontlets should be provided so located that the tanks could be drained, effecting a large saving in sugar.

It is not necessary to wash these tanks very often. The wash-out valve should be out of the control of the filter-press man. These tanks should not be encumbered with coils, but should have double bottoms, that they may be easily discharged and washed.

Perhaps the most important improvement this season was in the work of the double effect. The substitution of a larger sweet-water pump enabled it to concentrate all the juice and obviated the necessity of employing an open pan as last season.

Among other improvements in progress at Magnolia is the introduction of tile drainage. Lines of tile will be laid 30 feet apart, running from the front or river to the rear or marsh side of the plantation. The ditches will be dug by machinery; 3 -inch, 4 -inch, 5 -inch, and 6 -inch tiles will be used.

Governor Warmoth intends draining 1,000 acres of land. This drainage system will necessitate an expenditure of about $\$ 12,000$.

Aside from benefits to be derived from better drainage there will be important advantages gained from closing the ditches. The amount of tillable land will be increased about 70 acres; the large annual expenditure for keeping the ditches and quarter-drains open will be stopped ; the cultivation will be easier and less expensive, and the steam plows can be handled more adrantageously.

Should the work of this station be continued very interesting and valuable data can be obtained.

## ACCIDENTS TO THE MILLS.

The mills were stopped several times by small pieces of iron. Some of this iron passed through the shredder; other pieces must have been introduced after the cane left the shredder. It was evident that this iron was put on the carrier maliciously. This trouble finally culminated in the breaking of the shell of one of the rolls of the supplemental mill. A few changes were made in the disposition of the laborers detailed for work about the mills, and the rolling was finished without further accident than the breaking of three couplings.

The average extraction and consequently the yield of sugar would have been larger had the supplemental mill not have been damaged.

## THE BAGASSE-BURNER.

The burner, ${ }^{1}$ as improved last season, continued to work satisfactorily. The consumption of coal was materially reduced.

CONSUMPTION OF FUEL.
The amount of coal consumed was determined as follows :
The coal was weighed for half the season and the total consumption was based on the data obtained.

Pounds of coal consumed in trenty one days, 436,335; arerage of 20,778 pounds per day. From the time the fires were kindled until the work was entirely finished was forty-five days, hence the total consumption was $45 \times 20,778=935,010$ pounds. The total quantity of sugar made was $1,159,768$ pounds, therefore the consumption of coal per 1,000 pounds of sugar was 906 pounds or 4.42 barrels. These figures do not include the coal used in the revivification of the cliar.

This house could be worked almost entirely without coal if the follow. ing improvements were made:
(1) Substitute copper coils for the iron ones now in use in the clarifiers.
(2) Discontinue the practice of clarifying the juice after defecation.
(3) Introduce a condenser employing juice to condense the rapors from the double effect. An illustration of such apparatus is giren opposite page 11t, Chem. Bulletin No. 5, and is there termed a Calorisateur à contre-courant.
(4) Substitute a low-pressure racuum-pan for the one now in use.

Under these conditions the juice-heater described in last jear's ${ }^{1}$ report would not be necessare. It could probabls be conserted into a condenser.

The question of economical engines is of not so great importance, since the exhanst steam is employed for eraporation.

## COMPOSITION OF THE JUICE.

The chemical work this season at Magnolia Station was not begun until the 24 th of November. Although the work of the sugar-house nominally commenced November 7 th, but very little cane was rolled before the 13th. The analyses, therefore, show the composition of the juice for all but eleven dass of the season; in which time 2,113 tons of cane were rolled. The cane jielded $S$ pounds less sugar per ton than the next 2,000 tons, consequently it is fair to presume that it contained less sucrose.

Table I.-Composition of the raw juices.
When two analyses are given the same date the first was sampled in the morning, the seeo: din the afternoon.]


Résuméshowing the mean composition of the raw juice for the campaign of $1=86$.
[ November 24 to December 20.]

|  | Means. | Maxima. | Minima. |
| :---: | :---: | :---: | :---: |
| Specific gravity. | 1.0665 | 1. 0739 | 1.0604 |
| Degree Baumé.... | 9.00 | 9. 90 | 8. 20 |
| Uegree Brix (per cent. total solids) | 16. 20 | 17.88 | 14.76 |
| Sucrose.......................per pen | 13. 50 | 15.04 | 11. 67 |
| Reducing sugars(glucose, se.) .. do Albuminnids | . 616 | .91 .3000 | . 61 |
| Albuminoids, per cent. sucrose... | 1. 246 | 2. 210 | . 0625 |
| Coefficient of prity | 83. 33 | 89.47 | 77. 17 |

## COMPARISON OF RAW AND CLARIFIED JUICES.

In the following table analyses of the raw juices are only given for those days when the clarified juices were also examined:

Table II.-Comparison of raw and clarified juices.
[When two a:alyses are given the same date the first was sampled in the morning and the second in the afternoon.]

| Date. | Number. | Degree Baumé. | $\begin{aligned} & \text { Degree } \\ & \text { Brix. } \end{aligned}$ | Sucrose. | Reducing sugar. | Albuminoids per cent. sucrose. | Cocfficient of purity. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Nov. 25 | 3 | 2 | 16. 66 | Per cent. | Per cent. |  |  |
| 26 | 4 | 8.6 | 15.55 | 13.90 | 82 |  | 83. 43 |
| 26 | 5 | 9.1 | 16.37 | 13. 60 | 87 | 1. 32 | 78.97 |
| 27 | 6 | 8.6 | 15. 55 | 12. 00 | . 7.3 | 1.88 | 83.08 |
| 28 | 7 | 9.5 | 17.21 | 14.36 | . 46 | 1. 04 | 77. 17 83.43 |
| 28 | 8 | 9.5 | 17.07 | 14.21 | . 49 | 1. 14 | 83.24 |
| 29 | ! | 9.9 | 17. 88 | 15.04 | . 43 | 1.91 | 83.27 84.11 |
| 29 | 10 | 9.6 | 17.33 | 14.54 | . 53 | . 91 | 83.90 |
| 30 | 11 | 9.4 | 16. 90 | 13. 91 | . 46 | . 85 | 82. 30 |
| 30 | 12 | 9.2 | 16. 69 | 1421 | . 39 | . 8 | 85.14 |
| Dec. 1 | 13 | 9.5 | 17. 14 | 13.95 | . 68 | . 99 | 81.39 |
| 1 | 14 | 9.2 | 16. 63 | 13.97 | . 55 |  | 84.00 |
| 2 | 15 | 8. 9 | 16. 0 | 14.04 | . 6 ') | . 95 | 87.75 |
| 2 | 16 | 9.0 | 16. 29 | 13. 00 | . 58 | . 96 | 79.80 |
| 3 | 17 | 9.0 | 16. 23 | 13. 76 | . 69 | 1.00 | 84.78 |
| 3 | 18 | 9.0 | 16. 17 | 13.01 | . 77 | 1. 15 | 80.45 |
| 4 | 19 | 8. 6 | 15. 59 | 12.75 | . 49 | 1.42 | 81.78 |
| 4 | 20 | 8.8 | 15.78 | 13.12 | . 60 |  | 83.14 |
| 9 | 27 | 9.5 | 17.07 | 14.97 | . 72 |  | 87.69 |
| 10 | 28 | 8.6 | 15. 47 | 13.19 | . 48 |  | 85.26 |
| 10 | 29 | 8. 9 | 16. 11 | 13. 84 | . 45 | 1.49 | 8.5. 90 |
| 11 | 30 | 8. 9 | 16. 09 | 13.45 | . 46 | 1.49 | 83. 59 |
| 12 | 32 | 8.5 | 15. 39 | 12.32 | . 91 | 1. 22 | 80.5 |
| 12 | 33 | 8.6 | 15. 47 | 13. 56 | . 64 | 1.2- | 87.6 .3 |
| 13 | 34 | 8.5 | 15. 40 | 12.91 | . 72 | 1. 26 | 83.8; |
| 13 | 35 | 8.6 | 15.57 | 12.97 | . 51 | 2.31 | 83. 30 |
| 14 | 36 | 8.5 | 15.27 | 13.04 | . 51 | 1.92 | 85.39 |
| 14 | 37 | 8.8 | 15. 87 | 12.92 | . 72 |  | 81.41 |
| 15 | 38 | 9.1 | 16. 43 | 13.74 | . 55 |  | 83.62 |
| 15 | 39 | 9.0 | 16. 18 | 13. 85 | . 58 | 1. 26 | 85. 59 |
| 17 | 42 | 9.0 | 16. 19 | 13.76 | . 60 | 1. 18 | 84.99 |
| 17 | 43 | 9.2 | 16. 69 | 14.34 | . 52 | . 91 | 85.91 |
| 18 | 44 | 9.0 | 16. 29 | 13. 60 | . 55 | 1. 19 | 83.48 |
| Mea |  |  | 16. 26 | 13.58 | . 60 | 1. 24 | 83.52 |

Table II.-Comparison of raw and clarified juices-Continned.

| Date. | Number. | Degree Baımó. | Degree Brix. | Sucrose. | Reducing sugar. | Albuminoids per cent. sucrose. | Coefficient of purity. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Nov. 25 | 3 | 10.2 | 18.37 | Per cent. 14.92 | Per cent. |  | 81.22 |
| Nov. 26 | 4 | 9.9 | 17.87 | 14.85 | .79 | . 51 | 8.2.93 |
| 26 | 5 | 9.6 | $17.3 \ddagger$ | 14.38 | . 79 | . 53 | 82. 93 |
| 27 | 6 | 9.1 | 16.37 | 12.95 | 82 | 1. 01 | 79. 11 |
| 28 | 7 | 9.3 | 17.79 | 11.98 | . 61 | . 71 | 84.20 |
| 28 | 8 | 10.0 | 18.04 | 15.17 | . 50 | . 74 | 84.69 |
| 29 | 9 | 10.5 | 18.98 | 16. 01 | . 68 | .47 | 84.35 |
| 29 | 10 | 10.1 | 18.19 | 15. 71 | . 53 |  | 86.37 |
| 30 | 11 | 10.3 | 18. 70 | 16. 04 | . 61 | . 55 | 85.78 |
| 30 | 12 | 9.7 | 17.47 | 14.63 | . 46 | - | 83.74 |
| Dec. 1 | 13 | 9.7 | 17.46 | 14.11 | . 84 | . 71 | 80.81 |
| - 1 | 14 | 9.7 | 17. 39 | 14.59 | . 42 |  | 82.94 |
| 2 | 15 | 9.9 | 17. 87 | 14.27 | . 46 | . 88 | 79.85 |
| 2 | 16 | 9.4 | 16. 89 | 14.85 | . 55 | . 58 | 87. 92 |
| 3 | 17 | 9.6 | 17. 27 | 14. 56 | . 43 | . 69 | 84. 31 |
| 3 | 18 | 9.7 | 17. 47 | 14.86 | . 71 | . 59 | 85.06 |
| 4 | 19 | 9.9 | 17.79 | 14.64 | . 49 | . 56 | 82.29 |
| 4 | 20 | 9.4 | 16. 89 | 14. 18 | . 46 | ............. . | 83.96 |
| 9 | 27 | 9.1 | 16.40 | 14. 20 | . 60 |  | 86.59 |
| 10 | 28 | 9.5 | 17. 07 | 14. 57 | . 62 |  | 85.35 |
| 10 | 29 | 9.5 | 17. 19 | 15.07 | . 49 | . 87 | 87.67 |
| 11 | 30 | 9.6 | 17.39 | 15.15 | . 50 | . 83 | 87.11 |
| 12 | 32 | 9.5 | 17. 09 | 14.56 | . 70 | . 86 | 85.20 |
| 12 | 33 | 8.8 | 15. 87 | 13.48 | . 62 |  | 84.94 |
| 13 | 34 | 9.0 | 16. 31 | 14. 07 | . 65 | . 98 | 86.14 |
| 13 | 35 | 9.6 | 17. 39 | 15. 09 | . 63 | . 79 | 86.77 |
| 14 | 36 | 9.7 | 17.47 | 14.94 | . 50 | 1.09 | 85.52 |
| 14 | 37 | 9.4 | 16. 94 | 14. 31 | . 48 |  | 84.47 |
| 15 | 38 | 9.6 | 17. 33 | 14.91 | . 62 |  | 86.04 |
| 15 | 39 | 10.5 | 18. 93 | 16.35 | . 43 | . 95 | 86.37 |
| 17 | 42 | 10.1 | 18. 17 | 15.91 | . 45 | . 63 | 87.56 |
| 17 | 43 | 9.7 | 17.47 | 15.23 | . 55 | . 57 | 87.18 |
| 18 | 44 | 9.4 | 17.00 | 14.90 | . 64 | . 59 | 87.65 |
| Mea | ns. |  | 17.46 | 14.80 | . 59 | . 72 | 84.76 |

The increase in the coefficient of purity, as shown by Table II, was 1.24. There was but little change in the relative proportions of sucrose and reducing sugars. In the raw juice the average quantity of reduc ing sugars per cent. sucrose was 4.41 ; after clarification this proportion was reduced to 3.98 . This slight reduction was probably due to the formation of a glucosate of lime, which was subsequently decomposed, leaving the products of the decomposition in solution.

In order to render the percentages of albuminoids more readily comparable they have been expressed in terms of the sucrose. To obtain the actual per cent. albuminoids based on the weight of the juice, multiply the per cent. sucrose by the number given in the column albuminoids per cent. sucrose.

In the raw juice the albuminoids per cent. sucrose was 1.24 , and in the clarified juice .72 , showing that some of the total albuminoids were still retained by the juice.

In 1884 the processes of defecation and clarification removed 45.71 per cent. of the total albuminoids; in 1885-'86, 45.01; in 1886, 41.93 .

## rOMPARISON OF RAW, CLARIFIED, AND FILTERED JUICES AND FILTERED SIRUP.

Analyses were conducted for a period of sixteen days to determine the effect of filtration through animal charcoal on the juice. These analyses are given in Table III.

Table III.-Comparison of raw juices, clarified juices, filtered juices, and filtcred sirup for a period of sixteen days.

RAW JUICES.

| Date. | Number. | Degree Baumé. | Degree Brix. | Sucrose. | Reducing sugar's. | Coefficient of purity. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Nov. 25. | 3 | 9. 2 | 16. 66 | Percent. $\text { 13. } 90$ | Per cent. .82 | 83. 43 |
| - 26 | 4 | 8.6 | 15. 55 | 12. 28 | . 87 | 78.97 |
| 26 | 5 | 9.1 | 16.37 | 13. 60 | . 74 | 83.08 |
| 27. | 6 | 8. 6 | 15. 55 | 12. 00 | . 73 | 77.17 |
| 28 | 7 | 9.5 | 17. 21 | 14.36 | . 46 | 83.43 |
| 28. | 8 | 9.5 | 17.07 | 14.21 | . 49 | 83.24 |
| 29. | 9 | 9.9 | 17.88 | 15. 04 | . $4: 3$ | 84.11 |
| 29 | 10 | 9.6 | 17.33 | 14. 54 | . 53 | 83.90 |
| 30 | 11 | 9.4 | 16. 90 | 13.91 | . 46 | 8.3. 30 |
| 30 | 12 | 9.2 | 16. 69 | 14. 21 | . 39 | 85.14 |
| Dec. 1. | 13 | 9.5 | 17. 14 | 13.95 | . 68 | 81.39 |
| 2. | 15 | 8.9 | 16. 00 | 14.04 | . 60 | 87. 75 |
| 2. | 16 | 9.0 | 16. 29 | 13. 00 | . 58 | 79. 80 |
| 3. | 17 | 9. 0 | 16. 23 | 13. 76 | . 69 | 84.78 |
| 4. | 19 | 8.6 | 15. 59 | 12. 75 | . 49 | 81.78 |
|  | 20 | 8.8 | 15.78 | 13. 12 | . 60 | 83.14 |
| Means . |  | 9.1 | 16.52 | 13. 67 | . 60 | 82.75 |

CLARIFIED JUICES.


FILTERED JUICES.

| Nov. 25. | 3 | 9. 0 | 16. 17 | 12. 48 | 61 | 77.18 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 26............ | 4 | 9.8 | 17. 67 | 14.44 | 77 | 82. 86 |
| 26........... | 5 | 9. 2 | 16.61 | 13.71 | . 77 | 82. 54 |
| 27. | 6 | 9. 5 | 17. 10 | 14.25 | . 89 | 83. 51 |
| 28. | 7 | 9. 6 | 17.33 | 13.60 | . 87 | 78.48 |
| 28. | 8 | 10.6 | 19.14 | 16. 19 | . 610 | - 84.59 |
| 29. | 9 | 10.2 | 18.58 | 15. 34 | . 48 | 82.56 |
| 29. | 10 | 10.7 | 19. 29 | 16. 24 | . 55 | 84.19 |
| 30. | 11 | 12.3 | 29.29 | 18.52 | . 67 | 83. 35 |
| 30. | $1 \because$ | 10.2 | 18.37 | 15.97 | . 44 | 83.12 |
| Dec. 1. | 13 | 10.7 | 19. 39 | 15.63 | . 66 | 80.61 |
|  | 15 | 9.7 | 17.47 | 14.20 | . 51 | 80.28 |
| 2 | 16 | 9.4 | 16. 89 | 14.45 | 6 | 84.96 |
| 3 | 17 | 10.6 | 19.07 | 14. 9.5 | . 70 | 78.40 |
| 4............ | 19 | 9.5 | 17.19 | 14.56 | . 58 | 84.70 |
| 4 | 20 | 9.1 | 16.49 | 14.08 | . 44 | 85.26 |
| Means |  | 10.0 | 18.017 | 14.87 | . 64 | 82. 14 |

Table. 11I.-C'omparison of raw juices, clarified juices, fe.-Continued fleteled shides.

| Date. | Nommber. | Degreo Baumó. | Degree Brix. | Sucrose. | Reducing sugar's. | Comflicient of purity. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Nov. 25. | 3 | 24.3 | 44.55 | Per cent. 36. 10 | Per cent, 1. 88 | 81.07 |
| 26. | 4 | 23.4 | 42.79 | 35. 93 | 2.29 | 83.97 |
| 26. | 5 | 21.7 | 39.56 | 33.22 | 2.35 | 83.97 |
| 27. | 6 | 22.3 | 40.70 | 33.56 | 2.61 | 83. 46 |
| 28 | 7 | 23.3 | 42.59 | 34.84 | 3.00 | 81.80 |
| 28. | 8 | 22.6 | 41.36 | 33.01 | 2. 19 | 79.81 |
| 29. | 9 | 19.8 | 36.09 | 38.00 | 1. 67 | 7758 |
| 29. | - 10 | 23.6 | 43.15 | 36. 10 | 1.96 | 83. 66 |
| 30. | 11 | 23.4 | 42. 92 | 35.75 | 1.90 | 83.29 |
| 30. | 12 | 23.7 | 43. 40 | 36. 54 | 1. 88 | 83.96 |
| Dec. 1. | 13 | 23.6 | 43. 22 | 35.49 | 1. 67 | 82.11 |
| 2 | 15 | 21.4 | 39. 10 | 32.58 | 1.98 | 83.32 |
| 2 | 16 | 24.8 | 45.38 | 37. 23 | 1. 65 | 82.04 |
| 3 | 17 | 24.1 | 44. 10 | 36.24 | 1. 70 | 84.44 |
|  | 19 | 24.3 | 44. 60 | 36.91 | 1.48 | 82.76 |
|  | 20 | 20.8 | 38.00 | 31.80 | 1. 64 | 83.68 |
| Means |  | 23.0 | 41.17 | 34.58 | 2.00 | 82.39 |

The results of these analyses are not surprising, when we consider the quality of the bone-black used. Mr. O. B. Stillman, a Boston reiner, examined this char and pronounced it to be in a very bad condiion. It weighed nearly seventy pounds to the cubic foot. This is early twenty pounds heavier than good char should weigh.
The decolorizing properties of this bone-black were good, but as it vas already laden with impurities it did not improve the purity of the uice, but, on the contrary, reduced it. This was due to the impurities in the char being redissolved. The sirups being heavier, and already rearly saturated with soluble matter, yield their impurities more readily o the action of the bone-black and are improved in purity. ${ }^{1}$ Referance to the first report of this station, giving analyses of raw and filered juices, will show this same result. The char in use at that time vas even worse than last season.
I do not believe that the benefit from the mechanical filtration and he decolorization will balance the damage to the juice resulting from he use of spent bone-black.

THF: FILTER-PRESSES.
The use of the filter-presses was continued this season with even reater success than in 1885-'86.

[^0]Very few analyses of the press-cakes were made, except to determine their value for fertilizing purposes. Howerer, the few aualyses that were made are presented in the following table:

> Table IV.-Analyses of press-calie.

| Date. | Sucrose. |
| :---: | :---: |
| 1886. <br> November 30. | Per cent. 7.40 |
| December 1. | 7. 24 |
|  | 7.34 |
|  | 8.06 |
| Means ... | 7.51 |

No record was kept in the sugar-house of the quantity of juice recovered by the presses. In fact it would be quite difficult to determine just the proportion of the juice flowing from the presses that would be lost in ordinary work.

This season we found it necessary to empty one press once for every 2,889 gallons of juice expressed by the mills. The total amount of juice was $1,271,205$ gallons, hence the number of presses of press-cake was $1,271,205 \div 2,859=440$. The average weight of the press cake per press was 330 pounds : $330 \times 440=145,200$ pounds of press-cake for the entire season. The amount of press-cake per ton of cane was 20.15 pounds.
In the work with the ${ }^{1}$ experimental press in 1884 on thoroughly drained "blanket" scums the yield of juice was 80 per cent. The average skimmings and settlings, after long standing and decantation of the clear juice, are much thinner than the blanket, and would yield from 85 to 90 per cent. juice by filter-pressing. In order to under rather than over estimate the work of the presses, I will base my estimates on the actual yield obtained with the small experimental press. It may be well to state here that the work on a small scale was no better than with the large presses.
On the abuve basis the press-cake forms 20 per cent. of the total weight of skimmings filtered, hence $145,200 \div 20 \times 100=726,000$, the total number of pounds of skimmings. These figures show that eren on a low estimate $6 \frac{1}{2}$ per cent. of the juice is lost in the skimmings.

As the presses save at least 80 per cent, of this loss, the saving of juice at Magnolia this season was 580,000 pounds.

Is I have repeatedly stated, I consider this estimate a very low one. I have no doubt but that carefully conducted experiments would show a saving of 820,000 pounds of juice. With the filter-press arrangement at Magnolia it is impossible to keep is separate account of the juice recovered.
${ }^{1}$ Chem, Bulletin No. 5, p. 59.

## COMMERCIAL VALUE OF THE PRESS-CAKE.

The samples of press-cake sent to Washington for analysis reached there in such a damaged condition that it was useless to analyze them, hence the value of the cake is based on last season's determiuations. ${ }^{1}$

Total weight of press-cake 145,200 pounds, or 72.6 tons of 2,000 pounds each. The total value at $\$ 10.64$ per ton is $\$ 772.46$.

## PRESSURE-REGULATOR FOR FILTER-PRESSES.

In order to do good work with the presses it is essential to maintain a regular pressure, free from sudden changes or shocks. This can be accomplished by several methods, but the simplest and one as reliable as any other is the following, devised some years since by Mr. O. B. Stillman:

This apparatus cousists of a pump (Blake or other suitable pattern), an air-tight receiver similar to an ordinary montejus in shape, and a damper-regulator.

In the bottom of the receiver there is a central depression. The dis-charge-pipe from the pump connects directly with the receiver; the feedpipe to the presses passes through the top of the receiver and dips to the central depression ; connection with the damper-regulator is made by means of a small tube leading from the top of the receiver. The lever of the regulator is counected with the steam-inlet to the pump. The operation of the apparatus is as follows :

The pump is started and forces the liquor into the receiver, whence it passes through the feed-pipe to the presses. When the pressure exceeds that required the lever on the regulator raises and partly or entirely closes the steam-inlet to the pump, and as soon as the pressure falls below the limit the steam-valve is automatically opened.

The use of this apparatus relieves the press-man from the necessity of regulating the pump. The pressure increases or decreases gradually, hence the presses work to the best advantage. This device is employed at the Soledad Estate, near Cienfuegos, Cuba, and gives perfect satisfaction.

ANALYSES OF SUGARS.
Table V.-First sugars.

| Date. | Lot. | Sucrose. | Date. | Lot. | Sucrose. |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1886. |  | Per cent. | 1886. |  | Per cent. |
| Nov. ${ }_{24}$ | 14 | 99.3 | Dec. | 27 | 99.7 |
| 24 | 15 | 98.8 | 9 | 29 | 98.0 |
| 25 | 16 | 99.1 | 10 | 30 | 98.2 |
| 26 | 17 | 99.3 | 12 | 31 | 98.8 |
| 28 | 18 | 99.5 | 13 | 32 | 99.0 |
| 28 | 19 | 98.5 | 15 | 33 | 99.2 |
| 29 | 20 | 97.8 | 17 | 34 | 98.7 |
| 30 | 21 | 98.9 | 17 | 35 | 99.2 |
| Dec. ${ }^{1}$ | 22 | 98.9 | 19 | 36 | 98.9 |
|  | 23 | 98.8 | 20 | 37 | 98.8 |
| 3 | 24 | 97.8 | 21 | 38 | 98.9 |
| 4 | $\stackrel{25}{26}$ | 97.8 99.2 | Mean |  | 98.78 |

${ }^{1}$ Chem. Bulletin No, 11, p. 16,

Table VI.-Sicoond sugurs.


The first sugars graded "off white," the seconds were boiled to string proof. But two sugars were made.

ANALYSES OF MOLASSES.
Table VII.-Analyses of molasses from first sugars. (II Molasses.)

| Date. | $\stackrel{+3}{\substack{\text { + }}}$ |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | Per ct. | Per ct. | Per ct. | Per ct. |  |
| Nov. 25 | 16 | 30.9 | 57.08 | 35.00 | 5.88 | 30.66 | 5.15 | 61.32 |
| 26 | 17 | 29.9 | 55. 20 | 35.70 | 6. 25 | 32.34 | 5.66 | 64. 68 |
| 28 | 19 | 28.8 | 53.24 | 33. 00 | 6. 45 | 31. 02 | 6.06 | 62. 04 |
| 29 | 20 | 29.2 | 53.84 | 34.00 | 5. 00 | 31.57 | 4. 64 | 63.15 |
| 30 | 21 | 28.9 | 53. 30 | 34.20 | 4. 00 | 3 3. 08 | 3. 75 | 64. 16 |
| Dec. 1 | 22 | 30.3 | 56. 00 | 36.70 | 5.40 | 32.85 | 4.83 | 65.54 |
| 2 | 23 | 26.8 | 49.20 | 32.00 | 5.00 | 32.51 | 5.08 | 65. 02 |
| 3 | 24 | 29.6 | 54.70 | 35.00 | 6.37 | 31.99 | 5. 82 | 63.98 |
| 9 | 28 | 29.7 | 54.90 | 36. 20 | 5. 40 | 32. 96 | 4.92 | 65.92 |
| 12 | 31 | 29. 1 | 53.70 | 38.80 | 4. 63 | 36. 12 | 4.31 | 72.24 |
| 13 | 32 | 29.6 | 54.70 | 37.00 | 5.62 | 33. 82 | 5.14 | 67.63 |
| 15 | 33 | 29.1 | 53. 70 | 37.10 | 4.83 | 34.54 | 4.49 | 69.08 |
| 17 | 34 | 30.1 | 5.5. 50 | 39.80 | 4. 83 | 35.86 | 4.35 | 71.72 |
| 17 | 35 | 30.1 | 55.50 | 38.50 | 4.44 | 34. 69 | 4.00 | 69.38 |
| 19 | 36 | 29.5 | 54.50 | 38. 40 | 3.97 | 35.23 | 3. 64 | 70.46 |
| 20 | 37 | 28.7 | 52.80 | 38.70 | 5. 71 | 31.91 | 5.40 | 63.83 |
| Means |  | 29.4 | 54.24 | 35.94 | 5.24 | 33.13 | 4.83 | 66.25 |

Table VIII.-Analyses of molasses from second sugars. (III Molassts.)

| Date. | $\stackrel{3}{3}$ |  |  |  |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | Per ct. | ler ct. | Per ct. | Per ct. | Perct. | Perct. |  |
| Nov. 26 | 1 | 40.8 | 76. 92 | 32.40 | 39.90 | 19.75 | 32. 41 | 39.42 | 19.51 | 42. 64 |
| 26 | - | 40.3 | 75. 92 | 32. 30 | 40.17 | 19. 60 | 32.46 | 40. 20 | 19.67 | 42.71 |
| Dec. ${ }^{2}$ | 3 | 41.5 | 78.51 | 33.40 | 41. 21 | 13. 90 | 3.3. 23 | 39.78 | 13. 41 | 43.41 |
|  | 4 | 40.5 | 76.41 | 32. 80 | 40.39 | 13.15 | 32. 66 | 411. 21 | 13.03 | 42.97 |
| , | 5 | 40.5 | 76. 43 | 32.90 | 40.62 | 13. 90 | 32. 75 | 40.44 | 13. 84 | 43. 09 |
| 7 | - | 34. 8 | 64.85 | 2. 80 |  | 11.42 | 33.71 |  | 13.36 | 44.35 |
| 9 |  | 40.1 | 75.48 | 31. 20 | 39. 48 | 13. 80 | 31.30 | 39. 24 | 13. 94 | 41. 18 |
| 10 | 9 | 42.2 | 79.91 | 32. 50 | 40. 61 | 15. 74 | 31. 62 | 39.40 | 1.5. 31 | 41. 61 |
| 15 | 10 | 41.9 | 79. 31 | 33. 70 | 40. 68 | 14. 04 | 32. 27 | 38. 95 | 13. 44 | 42. 46 |
| 19 | 11 | 41.7 | 78.86 | 35.00 | 42.20 | 13. 80 | 33.78 | 40.72 | 13. 31 | 44.46 |
| Means |  | 40.5 | 76. 26 | 32.54 | 41. 13 | 1491 | 33.52 | 39.62 | 14.89 | 42.79 |

It will be noticed in the above tables that the analyses have been reduced to the basis of a molasses of a stated density. This facilitates the comparison of the results, the percentages being directly comparable with one another.
The arbitrary density $50^{\circ}$ Brix (Baumé 270.2 ) has been taken for molasses from firsts and $76^{\circ}$ Brix (Baumé $40^{\circ} .3$ ) for molasses from seconds.

Fifty and seventy-six degrees Brix were selected as standards of comparison, since they represent approximately the average densities of molasses from first and second sugars.

## LOSS OF SUGAR IN THE PROCESSES OF MANUFACTURE.

Unless careful arrangements for measurements are provided it is almost impossible to trace losses in manufacture. At Magnolia there is $n 0$ provision for measuring either sirups or masses cuites. The quantities of cane juice, sugars, and molasses are accurately known, and they, together with the analytical work, are the basis for the following calculations:

Loss of sugar in the processes of manufacture.
[Data for calculations.]
F'irst sugar (980.78 polarization), per ton of cane ..... pounds. . 120.47
Secoud sugar ( $90^{\circ} .30$ iolarization, .50 per cent. glucose) per ton of cane. do. ..... 40. 53
Molasses ( $41{ }^{\circ} .13$ double polarization, 14.91 per cent. glucose, 76.26 Brix), per. ton of cane pounds ..... 58.17
Press cake ( 70.51 polarization .30 per cont. glucose), per tou of cane... do. ..... 20.15
Sucrose in juice ..... 13.50
Glucose in juice .....  61
Juice per ton of cane ..... 1,560.00
Sucrose per ton of cane:
In first sugars ..... 119. 00
In second sugars ..... 36. 60
Iii press cakes ..... 1. 51
In molasses ..... 23. 93
Total sucrose accounted for as above ..... 181. 04
Total sucrose in juice extracted ..... 210.60
Sucrose to be accounted for ..... 29. 56
Sucrose in juice extracted ..... 210.60
Sucrose iu sugars and press cake ..... 157.11
Sucrose that would pass into the molasses were there no losises in manufact- uro ..... 53. 19
Glucose per ton of cane:
In the juice extracted pounds. ..... 9. 52
In the second sugars, . 20 pounds; press cake, 06 pounds do.... ..... 26
Glucose that would pass into the molasses were there no losses in man- ufacture ..... 9.36
19623-No. 15 ..... 2

Assuming that had there been no inversion and no losses in evaporation, \&c., the molasses would have contained the same proportion of non-sugars as the actual molasses did, we obtain the following:

|  | Theoretical composition of molasses. | Actual composi tion of molasses. |
| :---: | :---: | :---: |
| Sucrose........ .- per cent.. | 47. 77 | 41. 13 |
| Glucose...... . . . . . . . do. . | 8. 27 | 14.91 |
| Degree Brix. | 76. 26 | 76. 26 |
| Glucose, per cent. sucrose... | 17. 31 | 36.25 |
| Coefficient of purits......... | 62.64 | 53. 93 |
| Pounds molasses ............ | 111.97 | 58.17 |

The actual loss from inversion is the difference between the sucrose in the theoretical molasses and the actual: $47.77-41.13=6.64$ per cent. sucrose inverted, based on the weight of the molasses ; $58.17 \times .0664$ $=3.86$ pounds of sucrose per ton of cane.
The difficulty is to exactly locate this inversion. The analy ses of juices and sirups show rery little inversion between the juice box and the vacuum-pan, but the actual extent of this inversion could not be determined.

The beet-sugar manufacturers charge abont $.60^{l}$ per cent. the weight of the beets of sucrose to loss in the evaporation and bone-charcoal filters. This is based on an average sugar content in beet about the same as that in our cane. This corresponds to a loss of 12 pounds of sucrose.

> résumé.


Accounting for the sucrose on a basis of the weight of the juice we have:

Sucrose, per cent. juice.
First sugars. . . . . . .............................................................. . . pounds .. 7.628
Second sugars ........................................................................... 2.346
Molasses .................. .................. ......................................... . . do.... 1.5333
Press-cake .................................... ....................................... . do.... . . . 097
Inverted ......................................... . ........................................ . .248
Lost in evaporation, de .............................................................................. 68
Unaccounted for................................................................... . . 880

It will be seen from the above table that there is a large proportion of sugar unaccounted for, even after a liberal deduction for losses in evaporation, filtration, \&e. This loss of sugar should be located, if possible. Unfortunately the available data are insufficient to do so this season, and this important problem must be left to another crop.

## SUMMARY OF DATA COLLECTED AT MAGNOLIA STATION, SEASON OF 1886-'S7.

Governor Warmoth kindly allowed me free access to the records of the sugar-houses, from which the following data were obtained:

Table IX. -Tons of cane worked, weight of juice extracted, per cent. of extraction, weight of first and second sugars and molasses per ton of cane for each of the four periods ${ }^{1}$ into which the season was divided.

| Period. | $\begin{gathered} \text { Cane } \\ \text { worked. } \end{gathered}$ | Juice exiracted. | Extraction. | Weight of sugar per ton. |  | Molasses per ton. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  |  |  | First sugar. | Second sugar. |  |
| First perior | $\begin{aligned} & \text { Tons } \\ & 2,113.76 \end{aligned}$ | Pounds. <br> 3, 225, 385 | Per cent. 76. 29 | Pounds. 116. 40 | Pounds. 36. 27 | Pounds. 54.05 |
| Second period | 2, 410. 72 | 3,848, 740 | 79. 82 | 124. 6 | 43. ${ }^{3} 5$ | 63.12 |
| Third period | 920.47 | 1,430, 881 | 77. 72 | 111. 57 | 34. 16 | 48.53 |
| Fourth period | 1,758.35 | 2, 757, 870 | 78.42 | 124.43 | 44.59 | 61.06 |
| Total | 7, 203. 30 | 11, 262, 876 | Means: 78.17 | 120.47 | 40.53 | 58.17 |

${ }^{1}$ The divisions of the season into periods were arbitrary, and were made when bad weather or other cause of delay permitted a thorough cleaning of the house.

| Per cent. of yield, sugars | 8.05 |
| :---: | :---: |
| Per cent. of yield, first sugar | 6. 02 |
| Per cent. of yield, second sugar | 2.03 |
| Pounds first sugar (polarization $98^{\circ} .78$ ) per ton of | 120.47 |
| Pounds second sugar (polarization $90^{\circ} .03$ ) per ton of | 40. 53 |
| Total sugar per ton of cane | 161.00 |
| Per cent. of total sugar obtained in first product | 74.82 |
| Per cent. of total sugar obtained in second product | 25.18 |
| Total number of acres of caue rolled | 590.81 |
| Total tons of cane rolled | 7, 203.3 |
| 'Tons of. cane per acre | 12.19 |
| 'lotal pounds of sugar made. | 1,159,768 |
| Puouds of sugar per acre. | 1, 962.6 |
| Total ponnds molasses made (11.6 pounds per gallon) | 418,563 |
| Pounds molasses per acre. | 708.45 |
| Pouvds molasses per ton of cane | $58.1{ }^{7}$ |

Table X.-Comparison of yield of sugar and molasses, seasons of 1884-'85, 1885-'86, and 1836-'87.

|  | 1886-87. | 1885-'86. | 1884-'85. |
| :---: | :---: | :---: | :---: |
| Yield of first and sccond sugars ............... . per cent. . | 8.05 | 7.43 | 6.87 |
| Yield of all sugats ................................... do. |  |  | ${ }^{2} 7.92$ |
| Yield of first and second sugars per ton of cane. . ponnds.. | 161.00 | 148.75 | 137.39 |
| Yield of molasses per ton of cane ................... . do.... | 58.17 |  | 58.25 |

[^1]When in New Orleans in November I received an iuvitation to visit the experiment station and witness a test of the Kleemann process for the filtration of the juice. This test on a small scale was very successful, and demonstrated clearly that all the juice could easily be filtered through presses.
This process was invented by Fritz Kleemann, of Schoeningen, Brunswick, Germany. It has been patented in all sugar-producing countries.

The first tests on a large scale with cane were made in Demerara at the following estates: Nonpareil, Lusignau, Enmore, and others. The Nonpareil sugar-houses filtered the juice from 400 long tons of cane per day, through eight presses of eighteen chambers each. These tests were made in May, 1886, since when the process has been introduced into Porto Rico and Cuba.

The following description of the process was given me by Mr. Er'nst Schulze, representing the owners of the process :

The raw juice is treated in the defecators or clarifiers with lime, as usual, except that smaller quautity is required. The juice is then heated to a temperature between $160^{\circ}$ and $180^{\circ}$ Fahr., and finely ground lignite or brown coal is added. The lignite must be reduced to as fine a state of division as is practicable. The quantity of lignite to be added varies with the amount of sugar contained in the juice, and ranges from 5 to 10 per cent. of the weight of the sugar. The temperature of the mixture is maintained at from $1500^{\circ}$ to $170^{\circ}$ Fahr. fifteen or twenty minutes, and is then pumped to the filter-presses. The filtered juice passes directly to the evaporating pans, and the sirup without further clarification or settling can be immediately boiled in the vacuum-pan.

The juice left in the press-cake in the presses is obtained by displatement with cold water.

One thirty-chamber Kroog press will filter 20,000 gallons of juice treated by the Kleemann process in twenty-four hours, ample time being allowed for displacement of the juice left in the press-cake and for eleaning the presses, changing cloths, \&c. An ordinary laborer can manipulate the presses.
The amount of precipitate retained by a thirty-chamber press will average 770 pounds. This precipitate contains 50 per cent. of its weight of juice, nearly all of which can be obtained by displacement.
The amount of press-cake per ton of cane is approximately 46 pounds, half of which is juice containing an average of about 13 per cent. sucrose. Were this juice thrown away with the press-cake it would entail a loss of $46 \times .50 \times .13=3$ pounds sucrose per ton of cane worked.

Even at a low raluation per pound for sucrose, this loss, amounting to nearly 30,000 pounds of sugar for an average crop at Magnolia, is a
very large item. These figures are based on an extraction of 78 per cent. of juice from the cane.

The Kroog presses are proded with a graduated juice-box for measuring the anount of displaced juice. Practice has demonstrated that the theoretical amount of water required to displace the juice from the cakes is not sufficient, consequently a little more is always used. The quantity for a twenty-four-chamber press is $\mathbf{1 6 0}$ litres, and for a thirty-chamber press 200 litres of water.

Mr. Heine, of St. Burghard, near Halberstadt, Germany, made a number of experiments to determine the effect of using too much water. He analyzed the juice at different stages of the displacement process. The press used was a twenty-four chamber, Kroog system.

The normal juice from the presses had a coefficient of purity of 83.38 .
The results of these experiments are given in the following table. Uuder the column marked "litres" the quantity of displacement water is indicated:

| Litres. | $\begin{aligned} & \text { Degree } \\ & \text { Brix. } \end{aligned}$ | Snerose. | Coefficient of purity | Lime salts in 100 parts of dry matter in the juice, not precipitated by carbonic acid. Weighed as CaO. |
| :---: | :---: | :---: | :---: | :---: |
|  |  | Per cent. <br> 7.78 |  |  |
| 80 | 8. 80 | 7.78 713 | 84.29 84.87 | 0.9010 0.9095 |
| 120 | 7.83 | 6.41 | 81.89 | 0.98:0 |
| 160 | 7.00 | 5. 70 | 81. 43 | 1. 1265 |
| 200 | 5.50 | 4.15 | 75. 04 | 1. 6464 |
| 240 | 3. 10 | 2. 15 | 69.36 | 2. 9750 |
| 280 | 1. 83 | 1.04 | 56. 83 | 5. 1745 |
| 320 | 0.93 | 0.43 | 46. 23 | 8.7973 |
| 360 | 0. 80 | 0.31 | 38. 75 | 12.0902 |

The water employed in these experiments was cold. The reason for limiting the quantity of water to 160 litres per twenty-four-chamber press is evident from an inspection of the above table.

## KLEEMANN PROCESS.

In reply to a letter of inquiry, Mr. Kleemann, the inventor of this process, has kindly given me a detailed plan for working. The following extracts from his letter will be of interest to the planters:

The charcoal yon mention as having used I suppose was vegetable charcoal ground to a fine powder. * * * Certainly you have employed a substance that next to brown coal is the best aid to filtration, owing to its specific gravity and its absorbing powers of mucilaginous matters. Certain presantions, however, have to be arlopted in its use, and its price compares unfavorably with that of hrown coal. Even when close-woven filter-cloth is used in the press, fine partieles of char liave a tendency to pass through the cloth along with the juice for some time after filtration' has started, and this liquor has to be turned back to be filtered over again after a suffi-

[^2]ciently fine filtering zurface of foculent matters and charenal has been formed on the surface of the cloth.
Juice treated with brown coal, on the otherohand, runs quite bright or nearly so from the first rumning of the press, decolorizes considerably, and the cakes formedin the press part easier with their sugar when it is desired to wash them out than cakes formed of vegetable char.
For your futmre guidance allow me to suggest what I consider the best mode of treating juice with brown coal to effect rapid and good filtration and at the same time use the minimum amount of brown coal in the operations. I make this suggestion on the supposition you are carrying ont your experiments on an estate containing all the most morlern appliances for successfully treating after it has been filtered. namely, double or triple effects vacuum-pans and centrifugals.
After the juice has been defecated in the usual manner allow it to sulside for a few minntes, then brush off the skinmings into a tank and run the juice iuto another (preferably a circular one fitted with revolving stirrers), where the brown coal has to be added. As soon as the bottoms are reached they also have to be run into the skimmings tanks The brown coal is now added to the tank containing the juice, and after being thoroughly mixed it is forced through the filter press, finishing off with a pressure of about 75 pounds per square inch. In some cases 50 to 60 pounds pressure will be found quitesufficient, and a speed in the filtration ought to be obtained of 9 gallons per square foot of filtering area per hoor, or for a press containing 400 square feet of filtering area 3,600 gallons per hour of filtered liquor. After the thin juice has been brought to about 20 응 Bamé in the triple effect, certain albuminous compounds that were soluble in thin juice now ${ }^{1}$ precipitate out at the higher density, and it is necessary to again add some ascertained proportion of brown coal to the liquor and pass throngh a press before being taken into the racumepan.
By the action of the brown coal the liquor is considerably decolorized at this point, and as all gummy matters have been previously taken out of the juice the liquor will filter throngh rapidly by simple gravitation from a tank situated 6 to 10 feet above the press. The cakes taken from this press are used over again for the filtration of the defecated juice. The skimmings and bottoms are mixed either with fresh brown coal or with that taken from the gravitation-press and filtered in a press by themselves. By dividing the work in the manner I have suggested, more work can be got throngh in a givea time, with better results and using a smaller proportion of brown coal, than if it had been added to the juice, skimmings, and bottoms together in the defecating tank.

Tests were made on a laboratory scale with lignite from Arery's Island. The results were very satisfactors. It is proposed to repent these experiments on a large scale.

## TEST OF THE KLEEMANN PROCESS AT MAGNOLIA.

Early in December, at the request of Mr. D. D. Colcock, of the Sugar Exchange, New Orleans, the Commissioner of Agriculture directed me to test this process on a practical working scale.

A sufficient quantity of lignite could not be procured, so, in accordance with the suggestion of Mr. Ernst Schulze, representing the owners of the process, fiuely ground charcoal was sulstituted. Experiment

[^3]on a small seale showed that a slight modification of the process must be made where charcoal, bituminous coal, or certain other substitutes for lignite are employed.

The clarifiers at Magnolia are of the ordnary form and have a capacity of 533 gallons. The filter-presses were manufactured by the Hallesche Maschinenfabrik, of Balle, Germany. An ordinary pistonpump was used to force the juice through the presses. The juice was limed as usual, i. e., to neutrality. In order to determine the amount of charcoal required, experiments were made with varying quantities.
(1) Ten per cent. of the weight of the sugar in the juice.
(2) Seven and a half per cent.
(3) Five per cent.

Any difficulty in filtration would indicate too little charcoal. As a result of this experiment it was found that the juice filtered equally well with 5 per cent. as with 10 . Five per cent. is probably as little as could be successfully employed.
The juice was rapidly heated to the boiling point, after liming, before the addition of the charcoal. The charcoal having been added the mixture was boiled and stirred thoroughly for ten or fifteen minutes and then forced through the presses.

One twenty-one-chamber press filtered 2,670 gallons of juice in three hours, at the end of which time it was opened and the press-cake removed. The chambers of these presses are not as large as those of the Kroog presses.

The filtered juice was perfectly clear and bright. It was immediately converted into sirup in the donble effect. This sirup was as bright as the filtered juice. A portion of the sirup after standing sereral days in a glass vessel did not show the slightest sediment.

Analyses were made of the juice at frequent intervals during the work. A portion was taken from each sample for the determination of the albuminoids.

The proportion of albuminoids are expressed, in the table, both as a percentage of the weight of the juice and in terms of the sucrose.

The first sample of the juice was taken from the first clarifier; and the first sample of clarified juice from the first portion of the filtered juice, consequently these samples represent the same juice before and after clarification. The rest of the samples were taken at intervals from the presses and from every third clarifier of juice.

The average of these results will represent as nearly as possible the same juice before and after treatment.

Table XI.-Shouing analyses of juices before and after treatment by Eleemann process.


Arerage increase in coefficient of purity equals 2.14.
In the preceding table column $A$ represents raw juices, column $B$ juices treated by Kleemann process. Referring to the table we find the average increase in the coefficient of purity by the ordinary process to be 1.24. Table XI shows an increase of 2.14 by the Kleemann process. This large increase in the purity of the juice wonld give a decided increase in the yield of sugars.

## THE ALBUMINOIDS.

The reduction in the percentage of albuminoids ras not as large as by the ordinary process. By the Klecmann process an average of 35.17 per cent. of the albuminoids were removed; by the ordinary process the reduction was nearly 45 per cent. I do not know to what extent this difference in the albuminoids would affect the working of the sirup. The sugarmaker reported that the sirup made by the Kleemann process in this test worked as easily as by the ordinary.

## TIIE PRESS.CAKE.

The following analysis of the presseake shows its value as a fertil. izer:
Per cent. moistnre ..... 46. 0.
Per cent. phosphoric acid ( $\mathrm{P}_{2} \mathrm{O}_{5}$ ) ..... 1. 6.4
Per cent. nitrogen ..... 4.2
Pounts phosphoric acid per ton ..... 32.0
Value of plosphoric acid, at 9 eents per ponme ..... §2. 95
Poums nitrogen per ton. ..... 8.4
Value of titrogen, at 19 embts per poumd ..... \&1. 60
Total commerrial valur. per ton ..... 84. 5

This process yields twice as much press cake as the ordinary.

## ADVANTAGES OF THE KLEEMANN PROCESS.

The increased coefficient of purity is not the only advantage of this process. There is an increase in the yield of sugar due to rapidity of working both juice and sirup. The quantity of sugar lost in the scums is reduced to a minimum, and the expense for labor is less.

In addition the presseake is in an excellent mechanical condition for use as a fertilizer.
This process certainly merits a more thorough test both by the Department and the planters. Lignite of good quality, I am informed, is abundant near the sugar area of Louisiana, and can be obtained at a small cost.

## CUBAN SUGAR-HOUSES.

Although the average Cuban sugar-house does no better work than the average in Louisiana, there are many estates where the results obtained are as good, if not better, than those of this country.
The enormous size and the great strength of the milling machinery is very noticeable. The carts are backed up to the carriers npon which the cane is thrown without attempting to place each stalk lengthwise of the carrier, as is usual in Louisiana. The mills are fed very heavily. The average percentage of extraction of the juice is low. This is largely due to the woodiness of the cane. The proportion of mare or fiber is considerably higher than in Louisiana.

On leaving the mill the juice passes through a calorisator, which utilizes the waste heat from the double or tripple effect and condenses a large proportion of the vapors. This is a double economy. It reduces the quantity of water required for the house and effects a notable saving in consumption of fuel. Very few Cuban sugar-houses have a sufficient supply of water.

After learing the calorisator the juice is conducted into double-bottom pans of 750 gallons capacity, termed defecators. Here it is limerd and heated to the boiling point, then settled. The clear jnice is drawn off and the scum and settlings are run into "blow-ups." The contents of the "blow-ups," after heating to the boiling point, are filter-pressed as at Magnolia. The clear juice, including that coming from the filterpresses, is immediately concentrated.

Many of the houses still employ high-pressure vacum-pans. The best practice for obtaining a maximum yield of first sugar is to make a medium grain, boiling at as low a temperature as possible and discharge the masse cuite as stiff as the strike-valve of the pan will permit. The masse cuite is dropped into wagons and left several loours to become cold.

Through the courtesy of Mr. E. F. Atkins, of Soledad, I am enabled to present a tableshowing the work of his honse last season. This house
employs double-milling. The treatment of the juice is essentially that described above. The Soledad estate is the property of Messrs. E. Atkins \& Co, of Boston. It is located on the Cannao River, about 15 miles from Cienfuegos.

Monthly results of worli on Soledad estate for year 1836.

|  | $\underset{\underset{\sim}{3}}{\stackrel{\text { B }}{3}}$ |  | 范 | $\frac{\tilde{x}}{\frac{\pi}{4}}$ | 家 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Juice from cane. | 66. 79 | 66. 73 | 66.94 | 65. 77 | 65. 25 | 66.30 |
| Density of juice Baumé | 9. 40 | 10.17 | 10.70 | 11.09 | 11.19 | 10.49 |
| Polarization of juice... | 15. 29 | 16. 32 | 17.10 | 17.46 | 18. 20 | 16. 87 |
| Coefticient of purity of juice | 88.30 | 88.86 | 88.43 | 88.10 | 90. C0 | 88.74 |
| 'Iotal sugars in juice......... | 10. 21 | 10.89 | 11.44 | 11.48 | 11.87 | 11.18 |
| Yield of first sugars fiom cane.......... | 7.89 | 8.50 | 9.02 | 8.88 | 8.98 | 8.66 |
| Test of first sugars ... .................... | 93. 61 | 96. 70 | 96. 40 | 96. 60 | 96. 3 : | 9652 |
| Field of second sngars from cane | 1. 49 | 1. 53 | 1. 82 | 1.75 | 2. 08 | 1.74 |
| 'Test of second sugars ............ | 88.00 | 8.j. 60 | 86. 80 | 86.80 | ¢6. 87 | 87.01 |
| Total sugars from cane | 9.38 | 10.03 | 10.84 | 10.63 | 11.06 | 10. 40 |
| Test of tirst molasses . | 50.36 | 53. 63 | 54.10 | 5.3. 20 | 54.56 | 53. 17 |
| Test of second molasses : | 38.10 | 38.00 | 41.50 | 39. 70 | 40.43 | 39.55 |
| Yield of first sugars, calculated into pure sugar of 100 polarization | 7.62 | 8. 22 | 8. 70 | 8. 58 | 8.64 | 8.35 |
| Yield of second sugars, calculated into pure sngar of 100 polarization | 1.31 | 1. 32 | 1. 58 | 1. 50 | 1. 81 | 1. 50 |
| Total yiehd of sugars, calculated into pure sugar of 100 polarization | 8.93 | 9.54 | 10.28 | 10.08 | 10.45 | 9.85 |

Abore percentages are calculated upon 100 pounds of cane ground.
Mr. Atkins is conducting a series of experiments to determine what cane will give him a maximum output of sugar at the least cost. Such experimeuts if made in Louisiana would probably prore very valuable. The following table gives the results of analyses of cane from experimeutal plots :

Analyses of cancs from experimental plots.

| $\frac{\stackrel{\tilde{Q}}{\tilde{B}}}{\underset{y}{z}}$ |  | Date. | Per 100 lbs . cane. |  | Analysis of juice. |  |  |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | Variety. |  |  |  | $\frac{\dot{\infty}}{\dot{B}}$ |  | $\begin{aligned} & \text { as } \\ & 0 \\ & 0 \\ & 0 \\ & 0 \\ & 0 \\ & 0 \end{aligned}$ |  |  |  |  |
| 1 | Crystalina | Mar. 26 | 72.78 | 27. 22 | 20.9 | 11.6 | 91.8 | 19. 2 | . 66 | 1.04 | 13. 97 |
| 2 | Red Ribbon. | Mar. 26 | 73.17 | 26.83 | 20. 2 | 11.2 | 91.5 | 18.5 | . 14 | 1. 56 | 13.54 |
| 3 | .-..do. | Apr. ! | 73.17 | 26. $8: 3$ | 21.7 | 12.0 | 93.0 | 20. 2 | . 10 | 1. 40 | 14.78 |
| 4 | Crystalina | Apr. 9 | 71.58 | 28.42 | 2.2. 0 | 12. 2 | 81. 3 | 17.9 | 1. 80 | 2.30 | 12.81 |
| 5 | . . . do. | Apr. 17 | 73.07 | 26. 93 | 21.7 | 12. 0 | 89.8 | 1). 5 | . 13 | 2.07 | 14.2 .7 |
| 6 | Red Ribbon.. | Apr. 17 | 70.29 | 29. 71 | 20.4 | 11. 3 | 90.1 | 18. 4 | . 15 | 1. 85 | 12.93 |
| 7 | Crystalina .. | A M: 19 | 74.52 | 25. 4.9 | 2.20 | 12.5 | 90.7 | 20.5 | . 20 | 190 | 15. 27 |
| 8 | Rerl Ribbon.. | A 1 r. 19 | 76.23 | 23.77 | 22.3 | 12.3 | 91.7 | 20.5 | . 15 | 1. 65 | 15. 63 |
| 9 | Crystalina.. | Apr: 20 | 71.48 | 28.52 | 29.4 | 12. 4 | 90. 1 | 20.2 | . 28 | 1.92 | 14. 43 |
| 10 | Red Iibbon | A1120 | 71.85 | 28.15 | 21.9 | 12. 1 | 91.3 | 20.0 | . 31 | 1. 69 | 14. 37 |
| 11 | J3ack Java | Apr. 21 | 71.68 | 28.32 | 22.0 | 12. 2 | !)i. 8 | 21.3 | trace | . 70 | 15. 26 |
| 12 | .. do... | Apr. 24 | 71.64 | 28.36 | 21.4 | 11.8 | 96.3 | 20.6 | . 08 | . 71 | ${ }^{1} 14.75$ |

${ }^{1}$ Sucrose in juice from 100 pounds came.
Note.-Samples 1 and 2 from fiehs fonr years planted; samples $3,4,5,6,7,8,9$, ancì 10 , spring cane, ten months planted; samples 11 and 12 . known as caña obscura fiek. right yoars planted. Tbe samples 1 to 10 were fiom river lands, 11 to 12 from side hill.

The results of these and other experiments indicate that the Black Java is the largest sugar producer for this locality.

These experiments are under the charge of the chemist of the house, Mr. F. A. Terry, of 19 Exchange Place, Boston.

## CAROLINA ESTATE.

The work of the Carolina estate, near Cienfuegos, will be of great interest to Louisiana planters. Mr. John O'Bonrke, the administrator of this estate, has kindly given me a statement of last season's erop. This is the only house in Cuba employing an eight-roller mill.

## THE MILL.

This may be termed a combination mill, consisting of four pairs of rolls of uniform length and diameter, and driven at the same periphery speed. The rolls are so arranged in a horizontal plane that the cane passes from one pair to the next without the use of a carrier, a narrow knife only being employed.

Dimensions.


Periphery speed of rolls from 18 to 20 feet per minute. Maximum capacity of the mill, 30 tons of cane per hour.

The bagasse is saturated with live steam between the second and third, and third and fourth pairs of rolls.

The mill is double-geared, fifteen to one, $i . e$. , fifteen revolutions of the engine to one of the rolls. The motive power is a pair of engines 21 inch cylinders and 41 inch stroke. The steam for the house is furnished by six boilers 40 by 6 feet; two flues. The milling machinery was built by Brissonneau, Fréres \& $\mathrm{C}^{\text {ie }}$, of Nantes, France.

## TREATMENT OF THE JUICE.

The juice is defecated in the usnal manner. The clear juice is drawn off and is claritied ly boiling and skimming. The sirup is boiled, as at Magnolia, in a ligh-pressure pan. The masse cuite falls directly into the mixer and is centrifugaled hot. Two molasses sugars are made.

| Cane ground (e,000 pounds) | 33,750.87 |
| :---: | :---: |
| Juice extracted | 68.15 |
| Sugars extracted | 10. 633 |
| First sugar | 7.634 |
| Molasses sugar | 3. 10:3 |
| Sugar mado | 7,948,50\% |
|  | 212. 74 |

Unfortunately this house does not employ a chemist, consequently these figures are of but little value for comparison with those obtained at Magnolia.

The mangement of the Carolina estate should be congratulated upon the neatness and the sytematic working of the sugar-honse.

## CENTRAL SAN LINO.

Reports of a successful burner for bagasse led me to visit this estate.
The American planters are probably aware that the handling of the bagasse is a large item in the expenses of a Cuban estate. The bagasse is carted to the yard and spread out to dry. After drying it is carted back to the sugar-house and is deposited in front of the boilers. It requires a large force of firemen to handle this material.

At San Lino the green bagasse is handled in a different manner from the above, and is burned directly after leaving the mills. This method of burning bagasse is known as the "Sodal."
This invention consists in a plan for burning green bagasse or any other damp combustible. The combustion of the bagasse itself dries the fresh bagasse, which is fed to the fires from the end opposite to the grate bars. A very simple device is employed to caray the fuel to the fires. It requires ten minates to transport the bagasse the length of boilers, in which time the flames passing over it evaporate a large proportion of the moisture.

The apparatus for charging the furnaces is driven by an independent, engine and is perfectly automatic in its operations. This apparatus has reduced the number of laborers employed ly about 150 men. In addition, the honse is enabled to work in all kinds of weather now that it is not dependent upon the sun to dry its fuel. This process is especially applicable to large houses rolling thirty to forty or more tons of cane per hour.

Throngh courtesy of the owners, Messrs. Montalvo, and M. Boulan ger, chemist of this house, I have obtained the following statisties, showing arerage of several seasons' work of this honse.

Statistics of cropss at Śan Lino.

|  | Decomber. | January. | Fubruary. | March and $\Lambda$ pril. |
| :---: | :---: | :---: | :---: | :---: |
| Extraction, per cont | 71.50 | 70.90 | 70.29 | 69.80 |
| Baんumó ................ | 8.00 | 8.80 | 9.50 | 10.50 |
| 1rix | 14. 20 | 15. 50 | 16. 80 | 18.60 |
| Sucrose, per cont | 11. 50 | 13.02 | 15.09 | 16.95 |
| Glucoso, per cent | 1. 45 | 1. 10 | 0.70 | 0.65 |
| Cocfliciont of purity.. | 80.99 | 84.00 | S9. 82. | 90.04 |
| Glucose, por cent. sucrose | 12. 60 | 8.44 | 4. 62 | 3. 88 |
| Missse cuite, por cent. cano | 11. 09 | 12.34 | 13.40 | 13.99 |
| Yield of first sngar from masso cuite. | 61.00 | 65. 59 | 66.40 | 67.90 |
| Yield of first sugar, por cent. cano ... | 7.10 | 8. 10 | 8.90 | 9. 50 |
| Yield of second sugar, por cent. cano...... | 0.85 | 1.10 | 1.45 | 1.55 |
| Polarization of first sugar . . . . . . . . . . . . . . . . | 95.50 | 96.00 | 96.50 | 97.00 |

## CENTR 1 L SUGAR-IIOUSES.

The manufacture of sugar in Cuba is gradually being consolidated into large houses, the smaller estates furnishing the canc.

The Cuban manufacturers are begiming to realize what the Germans long since learned, viz, that it is more profitable for the sugar-house to attend to the manufacture only and leave the agriculture to tenants or independent farmers.

This system should be adopted by the planters of Louisiana. Our houses are too small to be worked economically. One house making $10,000,000$ pounds of sugar cau manufacture for a far smaller proportionate cost than three estates working three and a third millions each.

One of the owners of the largest houses in Cuba and probably the largest cane-sugar house in the world, told me that on the scale in which they manufacture ( 175,000 to 200,000 pounds per day) they make good profits even with the present prices. This house purchases its cane from tenants and independent farmers at a price dependent upon the average market value of sugar for each month. Hence, if the price of sugar rises, the cultivator profits by the rise; if it falls he divides the loss with the manufacturer. In these contracts, which are made for a term of years, it is specified that the density of the juice must not fall below a certain degree, but if it should, the manufacturer pay's a lower prices for the cane. The time of harvesting and the price to be paid for burnt cane in case of the fields being swept by tire also enter into these contracts.

## THE ECONOMY OF S'LEAM.

The cost of fuel is a very large expense on Cuban estates, especially if this fuel is coal, which must be imported from other comitries. The best houses employ every device for economy of combustible.

The evaporation is conducted in multiple-effect paus. The juice is pumperl from the mill to a calorisator, consisting of a large number of tubes about which the vapors from the laṣt pan of the system circulate,
and accomplishes a large proportion of the condensation. This is a rery considerable economy. In the final concentration in the strike-pan exhanst steam at about five jounds pressure is employed.

I believe that the work at Magnolia could be done with a very largely reduced quantity of coal, and possibly with bagasse and a very little wood only, if these improvements were adopted. This would not include the bone-black department.

# THE ANALSSIS OF SUGAR-CANE AND BEET'JUICES; WITII A TABLE FOR THE CALCULATION OF hNALISES. 

C. A. Crampton.

In the routine work of cane- and beet-sugar houses the greater part consists in the analysis of large numbers of samples of juice. A method of manipulation for these analyses is therefore desirable, which will unite a reasonable degree of accuracy with considerable speed. The method and rontine of amalysis given below has been pursued by myself and others for several years in the work of the United States Department of $A$ griculture in its experiments on sorghum and sugar-cane, and found very expedient in the brief "sugar season," when the saving of time is so important an object.

For very accurate work, of course it is necessary to weigh out the sample of juice takeu for analysis, but for all ordinary purposes the sample may be measured by volume, increasing greatly thereby the rapidity of the operation. By taking the specific gravity with accurate spindles, having their graduations well spaced, and correcting the volume taken, a very close approximation to the actual weight is obtained; this cor rectiou may be applied in two waye:
(1) By dividing the normal weight for the polariscope, or a multiple of it, by the specific gravity of the juice, to find the number of cubic centimeters of the liquid exactly corresponding to this weight, and then meas. uring out this quantity from a burette into a 100cc. flask, adding sub. acetate of lead solution, making up the volume to $100 c c$., filtering and polarizing. The reading of the polariscope gives directly the percentage of sugar in the juice, only multiplied by the number of times the normal weight was taken.
(2) By measuring out a certain volume of the juice, adding lead solution, making up to another definite volume, polarizing aud applying the correction for specific gravity to the reading obtained.

I prefer the latter method as more expeditious and convenient, for in the other the specific gravity must be determined and the calculation made before the sample for polarization can be taken; while in the second method the specific gravity may be taken at any time through the day, and the calculation of the results postponed indefinitels.

The custom annong sugar chemists in the beet honses in France and Germany is to measure 100 ce. of juice in a flask graduated to 100 and $110 c c$, add 10 ce. of a dilute solution of subacetate of lead, filter, and polarize; the increase in volume being corrected either by reading in it ?20mm. tube or by adding one teuti to the reading. I have found it more adrantageons, however, to take 50 cc . of the juice, add lead, and make the volume up to 100 cc . This gives a much more satisfactory liquid for polarization, especially with impure juices, and the filtration is greatly facilitated. The only object attained by using 100 cc . is the increase of the reading and consequent lessening of errors of observation. This is of but little importance, however, as 50ce. is three times the normal quantity for most instruments. Even this oljection may be done away with by using a double length ( 400 mm .) tube, providing the instrument used will admit of the long tube, and this is cutirely practicable so far as the lucility of the solution is concerned ; for I have never failed to get clear, bright solutions by proceeding in the manner I have indicated, except in cases where juices had been allowed to stand and mucous fermentation had set in.
The detailed procedure is as follows:
Fifty cubic centimeters of the juice is measured out by means of an accurately calibrated pipette and run into a 100 cc . flask; to it is added from 1 to tec. of a strong solution of sub acetate of lead,' the whole made up to the mark with water, filtered, and polarized.

The same solution is then used for the determination of the reducing sugars by Fehling's solution, the excess of lead being precipitated by sulphate of sola. As the reduction of copper sub-oxide by sugars has been shown to be affected by the amount of dilution in which the action takes place, it is desirable to have as nearly as possible a constant amount of dilution, and the most favorable dilution is such that from 10 to $20 c \mathrm{c}$. of the solution will be required to reduce the copper from 10 cc . of Fehling's solution; so, for juice presumed to contain from 1 to 2 per cent. of reducing sugars, $\check{50 c c}$. of the solution, which has been polarized, made up to 100 cc . will furnish the proper dilution; for those containing from 2 to 3 per cent., 50 to 100 cc ., and so on. In the case of juices containing less than 1 per cent. of reducing sugars the excess of lead may be removed by adding the sorke sulphate ${ }^{2}$ before the lead precipitate is tiltered off, when the same solution can be used for polarization and for the reducing-sugar determination. The copper solution used is Violette's modification, and the operation is performed in large test-tubes from 9 to 10 inches long and of about 1 to $1 \frac{1}{2}$-inch caliber, and the end reaction is obtained with ferrocyanide of potash and acetic acid, the small

[^4]portion of clear liquid required being filtered out by means of Wiley's tubes, already described. ${ }^{1}$ These tubes are simply glass tubes about 9 to 10 inches long and $\frac{1}{4}$-inch caliber. At one extremity a slight rim is made by turning back the edge while hot, and over this is stretched and tied a piece of fine linen cloth. To use the tube it is inserted into a beaker of water containing in suspension a quantity of very finely ground asbestos; by suction at the other end of the tube the linencap is covered with a film of asbestos and it is then immediately inserted into the test-tube containing the hot precipitated Fehling solution; suction is again made ou the tabe and the portion of clear liquid obtained is run into a porcelain dish containing the test solution by simply inverting the tube. These tubes may be cleaned and used over and over again by washing off the asbestos film, dissolving out any sub-oxide of copper which may remain in the cloth by soaking in nitric acid and rinsing in water until they no longer taste acid. One covering of linen suffices for many tests, and when worn out it is easily replaced.

Working in this way a large number of samples can be analyzed in a day, twenty-five to thirty being generally a fair day's work for operators on the stations of the Department of Agriculture, both sucrose and glucose being determined, and also the total solids (from the specific gravity), and this number is much exceeded in case of stress of work by allowing the polarizations to stand until night, using all the daylight for glucose determinations. By having a boy measure out the samples and prepare them for analysis the amount of work done can be nearly doubled.

It remains to calculate the results obtained, and to facilitate this calculation I have compiled the following table. In it will be found the specific gravity and degree Brix, ${ }^{2}$ and opposite to them factors for the calculation of sucrose and glucose in juices of the corresponding specific gravity, according to the method of procedure given above. Two factors for sucrose are given, one for instruments using as a norinal quantity 26.048 grams, such as Soleil-Scheibler, and one for those using 16.3 grams, such as the Laurent. These factors are the quotients from division of the normal quantity by $50 \times \mathrm{sp}$. gr. To use them it is necessary simply to multiply by the reading of the polariscope, the result giving directly the per cent. of sucrose in the juice. The glucose factor is simply the reciprocal of the specific gravity multiplied by ten. To use it, it is divided by the number of cubic centimeters of the normal solution ( 50 to 100 cc .) required to precipitate 10 cc . of Fehling's solution, which gives the result directly in per cent. If the solution has been diluted twice ( 50 to 200 cc .), divide the factor by half the number of cubic centimeters required; if one and one-half times ( 50 to 150 cc .), by one and one half times the number of cubic centimeters required.

[^5]Table for the calculation of sugar juices.

| Specific gravity. | Degree Brix. | Sucrose factor for instruments requiring- |  | Glucose factor. | Specific gravits. | Degree Brix. | Sucrose factor forinstruments requiring- |  | Glucose factor. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 26. 048 grams. | $\begin{gathered} 16.3 \\ \text { grams. } \end{gathered}$ |  |  |  | 26.048 grams. | $\begin{gathered} 16.3 \\ \text { grams. } \end{gathered}$ |  |
| 1. 0197 | 5. 0 | . 5109 | . 3197 | 9. 807 | 1. 0502 | 12.4 | . 4961 | . 3104 | 9. 523 |
| 1. 0201 | 5.1 | . 5107 | . 3196 | 9. 803 | 1.0506 | 12.5 | . 4959 | . 3103 | 9.519 |
| 1. 0205 | 5. 2 | . 5105 | . 3195 | 9. 799 | 1. 0510 | 12.6 | . 4957 | . 3102 | 9.515 |
| 1. 0209 | 5.3 | . 5103 | . 3193 | 9. 795 | 1. 0514 | 12.7 | .49.55 | . 3101 | 9.511 |
| 1. 0213 | 5.4 | . 5101 | . 3192 | 9.791 | 1. 0519 | 12.8 | . 4953 | . 3099 | 9. 507 |
| 1. 0217 | 5.5 | . 5099 | . 3191 | 9.788 | 1. 0523 | 12.9 | . 4951 | . 3098 | 9.503 |
| 1. 0221 | 5. 6 | . 5097 | . 3190 | 9.784 | 1. 0527 | 13.0 | . 4949 | . 3097 | 9. 499 |
| 1. 0225 | 5.7 | . 5095 | . 3189 | 9. 780 | 1. 0531 | 13.1 | . 4947 | . 3096 | 9.496 |
| 1. 0229 | 5.8 | . 5093 | . 3187 | 9. 776 | 1.0536 | 13.2 | . 4945 | . 3094 | 9.491 |
| 1. 0233 | 5.9 | . 5091 | . 3186 | 9. 772 | 1. 0540 | 13.3 | . 4943 | . 3093 | 9.488 |
| 1.0237 | 6.0 | . 5089 | . 3185 | 9.768 | 1. 0544 | 13.4 | . 4941 | . 3092 | 9. 484 |
| 1.0241 | 6.1 | . 5087 | . 3184 | 9. 765 | 1.0548 | 13.5 | . 4939 | . 3091 | 9.480 |
| 1.0245 | 6.2 | . 5085 | . 3183 | 9.761 | 1.0553 | 13.6 | . 4937 | . 3090 | 9.476 |
| 1. 0249 | 6.3 | . 5083 | . 3182 | 9.757 | 1. 0557 | 13.7 | . 4935 | . 3088 | 9.472 |
| 1. 0253 | 6.4 | . 5081 | . 3181 | 9.753 | 1. 0561 | 13.8 | . 4933 | . 3087 | 9.469 |
| 1. 0257 | 6.5 | . 5079 | . 3180 | 9.749 | 1.0566 | 13.9 | . 4931 | . 3085 | 9. 464 |
| 1:0261 | 6.6 | . 5077 | . 3178 | 9.746 | 1. 0570 | 14.0 | . 4929 | . 3084 | 9.461 |
| 1. 0265 | 6.7 | . 5075 | . 3177 | 9. 742 | 1. 0574 | 14.1 | . 4927 | . 3083 | 9.457 |
| 1. 0269 | 6.8 | . 5073 | . 3175 | 9.738 | 1.0578 | 14.2 | . 4925 | . 3082 | 9.453 |
| 1.0273 | 6.9 | . 5071 | . 3174 | 9. 734 | 1.0583 | 14.3 | . 4923 | . 3080 | 9.449 |
| 1.0277 | 7.0 | . 5069 | . 3173 | 9.730 | 1. 0587 | 14.4 | . 4921 | . 3079 | 9.446 |
| 1.0281 | 7.1 | . 5067 | . 3171 | 9.727 | 1. 0591 | 14.5 | . 4919 | . 3078 | 9.442 |
| 1. 0286 | 7.2 | . 5065 | . 3170 | 9.723 | 1.0596 | 14.6 | . 4917 | . 3077 | 9.438 |
| 1. 0290 | 7.3 | . 5063 | . 3169 | 9.719 | 1. 0600 | 14.7 | . 4915 | . 3076 | 9.434 |
| 1. 0294 | 7.4 | . 5061 | . 3167 | 9.715 | 1. 0604 | 14.8 | . 4913 | . 3074 | 9.430 |
| 1. 0298 | 7.5 | . 5059 | . 3166 | 9.711 | 1.0609 | 14.9 | . 4911 | . 3073 | 9. 426 |
| 1.0302 | 7. 6 | . 5057 | . 3164 | 9. 707 | 1. 0613 | 15.0 | . 4909 | . 3072 | 9. 422 |
| 1. 0306 | 7.7 | . 5055 | . 3163 | 9.703 | 1.0617 | 15.1 | . 4907 | . 3071 | 9.419 |
| 1. 0310 | 7.8 | . 5053 | . 3162 | 9.699 | 1. 0621 | 15.2 | . 4905 | . 3069 | 9.415 |
| 1. 0314 | 7.9 | . 5051 | . 3160 | 9.696 | 1. 0626 | 15.3 | . 4903 | . 3068 | 9.411 |
| 1. 0318 | 8.0 | . 5049 | . 3159 | 9.692 | 10630 | 15.4 | . 4901 | . 3067 | 9. 407 |
| 1. 0322 | 8.1 | . 5047 | . 3158 | 9.688 | 1. 0634 | 15.5 | . 4899 | . 3066 | 9.404 |
| 1. 0327 | 8.2 | . 5045 | . 3157 | 9. 684 | 1. 0639 | 15.6 | . 4897 | . 3064 | 9.399 |
| 1. 0331 | 8.3 | . 5043 | . 3156 | 9. 680 | 1. 0643 | 15.7 | . 4895 | . 3063 | 9. 396 |
| 1. 0335 | 8.4 | . 5041 | . 3154 | 9. 676 | 1.0647 | 15. 8 | . 4893 | . 3062 | 9.392 |
| 1. 0339 | 8.5 | . 5039 | . 3153 | 9.672 | 1. 0652 | 15.9 | . 4891 | . 3061 | 9. 388 |
| 1.0343 | 8.6 | . 5037 | . 3152 | 9. 668 | 1. 0656 | 16.0 | . 4889 | . 3059 | 9.385 |
| 1.0347 | 8.7 | . 5035 | . 3151 | 9. 665 | 1. 0660 | 16. 1 | . 4887 | . 3058 | 9. 381 |
| 1.0351 | 8.8 | . 5633 | . 3150 | 9. 661 | 1. 0665 | 16.2 | . 4885 | . 3057 | 9.37\% |
| 1. 0335 | 8.9 | . 5031 | . 3148 | 9. 657 | 1. 0669 | 16.3 | . 4883 | . 3056 | 9.373 |
| 1. 0359 | 9.0 | . 5029 | . 3147 | 9.653 | 1. 0674 | 16.4 | . 4881 | . 3054 | 9.369 |
| 1.036t | 9.1 | . 5027 | . 3146 | 9. 649 | 1. 0678 | 16.5 | . 4879 | . 3053 | 9.365 |
| 1. 0368 | 9.2 | . 5025 | . 3144 | 9. 645 | 1. 0682 | 16.6 | . 4877 | . 3052 | 9.362 |
| 1. 0372 | 9.3 | . 5023 | . 3143 | 9.641 | 1.0687 | 10.7 | . 4875 | . 3051 | 9. 358 |
| 1. 0376 | 9.4 | . 5021 | . 3142 | 9. 638 | 1. 0699 | 16.8 | . 4873 | . 3049 | 9.354 |
| 1.0380 | 9.5 | . 5019 | . 3141 | 9. 634 | 1.0695 | 16.9 | . 4871 | . 3048 | 9.350 |
| 1. 0384 | 9.6 | . 5017 | . 3140 | 9. 630 | 1.0700 | 17.0 | . 4869 | . 3047 | 9. 346 |
| 1. 0388 | 9.7 | . 5015 | . 3138 | 9. 626 | 1.0704 | 17.1 | . 4867 | . 3046 | 9.342 |
| 1. 0393 | 9.8 | . 5013 | . 3137 | 9. 622 | 1. 0709 | 17.2 | . 4865 | . 3044 | 9.338 |
| 1. 0397 | 9.9 | . 5011 | . 3136 | 9. 618 | 1.0713 | 17.3 | . 4863 | . 3043 | 9. 334 |
| 1.0401 | 10.0 | . 5009 | . 3134 | 9. 614 | 1.0717 | 17.4 | . 4861 | . 3042 | 9.331 |
| 1. 0405 | 10.1 | . 5007 | . 3133 | 9. 611 | 1. 0722 | 17.5 | . 4859 | . 3041 | 9. 327 |
| 1.0409 | 10.2 | . 5005 | .3132 | 9. 607 | 1.0726 | 17.6 | . 4857 | . 3039 | 9. 323 |
| 1.0413 | 10.3 | . 5003 | . 3131 | 9. 603 | 1.0730 | 17.7 | . 4855 | . 3038 | 9.320 |
| 1. 0418 | 10.4 | . 5001 | . 3129 | 9. 599 | 1.0735 | 17.8 | 1853 | . 3037 | 9.315 |
| 1.0422 | 10.5 | . 4999 | . 3128 | 9. 595 | 1.0739 | 17.9 | . 4851 | . 3036 | 9.312 |
| 1.0426 | 10.6 | . 4997 | . 3127 | 9. 591 | 1. 0744 | 18.0 | . 4849 | . 3034 | 9. 308 |
| 1. 0430 | 10.7 | . 4995 | . 3126 | 9. 588 | 1.0748 | 18.1 | . 4847 | . 3033 | 9. 304 |
| 1. 0434 | 10.8 | . 4993 | . 3125 | 9. 584 | 1. 0753 | 18.2 | . 4845 | . 3032 | 9. 300 |
| 1.0439 | 10.9 | . 4991 | . 3123 | 9. 580 | 1.0757 | 18.3 | . 4843. | . 3031 | 9.296 |
| 1. 0443 | 11.0 | . 4989 | . 3122 | 9. 576 | 1.0761 | 18.4 | . 4841 | . 3029 | 9.293 |
| 1. 0447 | 11.1 | . 4987 | . 3121 | 9. 572 | 1.0766 | 18.5 | . 4839 | . 3028 | 9. 289 |
| 1.0451 | 11.2 | . 4985 | . 3119 | 9. 568 | 1.0770 | 18.6 | . 4837 | . 3027 | 9.285 |
| 1. 6455 | 11.3 | . 4983 | . 3118 | 9. 565 | 1.0775 | 18.7 | . 4835 | . 3026 | 9. 281 |
| 1. 0459 | 11.4 | . 4981 | . 3117 | 9.561 | 1. 0779 | 18.8 | . 4833 | . $30 \geq 4$ | 9. 277 |
| 1.0464 | 11.5 | . 4979 | . 3115 | 9. 55.7 | 1.0783 | 18.9 | . 4831 | . 3023 | 9.274 |
| 1.0468 | 11.6 | . 4977 | . 3114 | 9. 553 | 1.0788 | 19.0 | . 4829 | . 3022 | 9. 270 |
| 1.0472 | 11.7 | . 4975 | . 3113 | 9. 549 | 1. 0732 | 19.1 | . 4827 | . 3021 | 9. 266 |
| 1. 0476 | 11.8 | . 4973 | . 3112 | 9. 545 | 1.0797 | 19.2 | . 4825 | . 3019 | 9. 262 |
| 1. 0481 | 11.9 | . 4971 | . 3110 | 9. 54 i | 1. 0801 | 19.3 | . 4823 | . 3018 | 9.258 |
| 1. 0485 | 12.0 | . 4969 | . 3109 | 9. 537 | 1.0806 | 19.4 | . 4821 | . 3017 | 9.254 |
| 1. 0489 | 12.1 | . 4967 | . 3108 | 9. 534 | 1.0810 | 19.5 | . 4819 | . 3016 | 9. 251 |
| 1. 0493 | 12.2 | . 4965 | . 3107 | 9. 530 | 1.0815 | 19.6 | .4817 | . 3014 | 9. 246 |
| 1. 0497 | 12.3 | . 4963 | . 3106 | 9.527 | 1.0819 | 19.7 | . 4815 | . 3013 | 9.243 |

Table for the calculation of sugar juices-Continued.

| Specific gravity. | $\begin{aligned} & \text { Degree } \\ & \text { Brix. } \end{aligned}$ | Sucros for inst requi | factor uments ing- | Glucose factor. | Specific gravity. | Degree Brix. | Sucrose factor for instrmments requiring- |  | Glucose factor. |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | 26. 048 grams. | 16.3 grams. |  |  |  | 26. 048 grams. | 16.3 grams. |  |
| 1.0824 | 19.8 | . 4813 | . 3012 | 9.239 | 1. 0914 | 21.8 | . 4773 | . 2987 | 9.163 |
| 1.0828 | 19.9 | . 4811 | . 3011 | 9.235 | 1. 0918 | 21.9 | . 4772 | . 2986 | 9.159 |
| 1.0832 | 20.0 | . 4809 | . 3009 | 9. 232 | 1. 0923 | 22.0 | . 4769 | . 2984 | 9.155 |
| 1.0837 | 20.1 | . 4807 | . 3008 | 9.228 | 1. 0927 | 22.1 | . 4768 | . 2983 | 9.151 |
| 1.0841 | 20.2 | . 4805 | . 3007 | 9.224 | 1. 0932 | 22.2 | . 4765 | . 2982 | 9.148 |
| 1. 0846 | 20.3 | . 4803 | . 3006 | 9. 220 | 1. 0936 | 22.3 | . 4764 | . 2981 | 9.144 |
| 1. 0850 | 20.4 | . 4801 | . 3005 | 9.216 | 1. 0941 | 22.4 | . 4762 | . 2980 | 9.140 |
| 1.0855 | 20.5 | . 4798 | . 3003 | 9.212 | 1. 0945 | 22.5 | . 4760 | . 2979 | 9.137 |
| 1.0859 | 20.6 | . 4797 | . 3002 | 9.209 | 1. 0950 | 22.6 | . 4758 | . 2977 | 9.132 |
| 1. 0864 | 20.7 | . 4795 | . 3001 | 9.205 | 1. 0954 | 22.7 | . 4756 | . 2976 | 9.129 |
| 1. 0868 | 20.8 | . 4794 | . 3000 | 9.201 | 1. 0959 | 22.8 | . 4754 | . 2974 | 9.125 |
| 1. 0873 | 20.9 | . 4791 | . 2998 | 9.197 | 1. 0964 | 22.9 | . 4752 | . 2973 | 9.121 |
| 1.0877 | 21.0 | . 4790 | . 2997 | 9.194 | 1. 0968 | 23.0 | . 4750 | . 2972 | 9.117 |
| 1.0882 | 21.1 | . 4787 | . 2996 | 9. 190 | 1. 0973 | 23.1 | . 4748 | . 2971 | 9.113 |
| 1.0886 | 21.2 | . 4786 | . 2995 | 9.186 | 1. 0977 | 23.2 | . 4746 | . 2970 | 9.110 |
| 1.0891 | 21.3 | . 4783 | . 2993 | 9.182 | 1. 0982 | 23.3 | . 4744 | . 2968 | 9.106 |
| 1.0895 | 21.4 | . 4782 | . 2992 | 9.179 | 1. 0986 | 23.4 | . 4742 | . 2967 | 9.102 |
| 1. 0900 | 21.5 | . 4779 | . 2991 | 9.174 | 1.0991 | 23.5 | . 4740 | . 2966 | 9.098 |
| 1.0904 | 21.6 | . 4778 | . 2990 | 9.171 | 1.0996 | 23.6 | . 4738 | . 2965 | 9. 094 |
| 1.0909 | 21.7 | . 4775 | . 2988 | 9.167 | 1.1100 | 23.7 | . 4736 | . 2963 | 9.091 |


[^0]:    ${ }^{1}$ Chem. Bulletin No. 5, pp. 49-50.

[^1]:    ${ }^{2}$ In 1884-'85 three sugars were made.

[^2]:    ${ }^{1}$ This diffienlty was net met with in the experiment at Magnolia. (x. L. S.

[^3]:    ${ }^{1}$ This was not the case in the experiment at Magnolia. The sirup remained perfectly clear withont the slightest trace of precipitate. There was no necessity for a subsequent settling or filtration of the sirup. G. L. S.

[^4]:    ${ }^{1}$ I'repared by rubbing up in a mortar 1 kilo of acetate of lead and $\frac{1}{2}$ kilo of litharge and hoiling with $\frac{1}{3}$ litre of water.
    ${ }^{2}$ For very exact work it is better to take a separate sample of the juice for the eopper test and dispense with the lead altogether if the purity of the juice admits, for the presence of lead produces an error, and even when precipitated by sulphate of sodat there is said to be an crror from the slight solubility of the sulphate of lead.

[^5]:    ${ }^{1}$ Bull. de l'Assoc. des Chim. de Sucre. et de Dist, Apr., 1884.
    ${ }^{2}$ Taken from the tables of Mategezek, Scheibler, and Stammer, as given by Dr. Tucker in his "Manual of Sugar Analysis."

    19623-No. 15-3

